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(54) **Flame retardant polyamide composition**

(57) Flame retarded resin molding compositions with enhanced electrical properties comprising a polyamide resin, a flame retarding amount of a halogenated

flame retardant, a fiber glass and/or mineral filler, and an effective amount of a pyrolypolyphosphate, and optionally including reinforcing glass fibers or an olefinic polymer for enhancing electrical properties.

## Description

This invention relates to flame-retarded resin molding compositions having improved electrical properties.

### Background of the Invention

Resins such as polyamide-6-6 are used in molding compositions due to many excellent properties, i.e. strength, solvent resistance, etc. In applications which require good electrical properties, molded parts may be exposed to severe conditions, such as exposure to electrical discharge or to excessive current leakage across its surface. These conditions require not only that the molded part have flame resistance, but also that the molded part have a high degree of resistance to carbonization upon exposure to high voltage, i.e. good track resistance and a high degree of resistance to electrical discharge.

However, it has been a problem that flame-retarded polyamide resins have poor track resistance compared to the non-flame-retarded resin. US Pat. No. 4,559,372 to Giroud-Abel describes a composition of comprising (i) a polyamide, (ii) a fireproofing amount of red phosphorus, and (iii) an effective amount of at least one anthranide compound as having improved electrical properties. Other materials mentioned include the incorporation of cupric oxide and cadmium oxide for improved arc resistance. This focus of this patent is on the utilization of red phosphorus.

Hence, it is desirable to provide additional ingredients which can enhance the track resistance of glass fiber polyamide composition, especially flame retardant polyamide compositions.

### Summary of the Invention

It is an object of the present invention to provide flame-retarded, reinforced polyamide resin molding compositions, likewise, articles molded from them which have improved electrical properties.

Another object of the present invention is to provide flame-retarded polyamide resin molded articles and the molding compositions which are suitable for molded electrical devices, such as electrical outlets, circuit breakers, switches, fuse holders, lamp sockets, distributor caps, or ignition coils.

According to the present invention, there are provided flame-retarded resin molding compositions with enhanced electrical properties, particularly track resistance, which comprise a polyamide, a flame-retarding amount of a halogenated flame retardant, fiber glass and/or mineral filler, and an effective amount of a pyro-polyphosphate for improving the electrical properties. The pyro-polyphosphate may be a metal pyrophosphate, a metal polyphosphate, a metal acid polyphosphate, or metal acid polyphosphate.

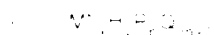
In accordance with another aspect of the present invention, an olefinic polymer is included in the composition, in addition to the pyro-polyphosphate for enhancing the electrical properties.

### Description of the Preferred Embodiments

In accordance with the principles of the present invention, the combination of ingredients are selected so that the compositions are suitable for electrical devices and have improved Comparative Tracking Index (CTI). CTI is typically measured by UL-746A or ASTM-D3638-85 or IEC-112-3rd Publication. The method of this test is intended to indicate the relative behavior of solid electrical insulating materials with regard to their susceptibility to surface tracking when exposed under electrical stress, to water, and other contaminants from the surrounding atmosphere. This method is especially suitable for testing synthetic resin moldings. Comparative Tracking Index (CTI) value is determined under the conditions specified in this test method, which will cause failure with the application of a 60 Hz power line electrolyte to the specimen, is used as a measure of the susceptibility of the material to the aging.

Preferably the CTI of the final composition is greater than or equal to that of the resin and preferably greater than about 400 volts. Typically, neat polyamide resin has a CTI value of about 100 volts. However, the addition of flame retardants and glass fibers tend to decrease the CTI value to about 275 volts or less. Hence, it is desirable to select the remaining ingredients so as to not detract from the desirable flame retardancy and mechanical properties but enhance the CTI of the final composition.

Preferably, the desired flame retardancy is V-0 according to Underwriters Laboratories Test No. 94. In addition to high CTI and flame retardancy, important properties include good mechanical properties, good processability, i.e. short molding cycle times, good flow, and good insulation properties. Flame retardant and high strength materials such as polyamide-6-6, polyamide-6, polyamide-6-12, polyamide-11, polyamide-12, polyamide-12-12, and polyamide-10-10 are suitable for this purpose.



wherein M is a metal, x is a number from 1 to 12, y is a number from 0 to 12, z is a number from 2 to 10, z is a number from 1 to 5 and the sum of (x+z)y is equal to or less than 12. M is preferably a Group I, A, IA, IB, or IIB metal and more preferably sodium or potassium.

These compounds include, for example, pyrophosphates of the form  $(Na_2H_2P_2O_7)_x \cdot (K_2H_2P_2O_7)_y \cdot Na_2H_2P_2O_7 \cdot zH_2O$ ,  $xNaH_2P_2O_7$  and  $Na_2H_2P_2O_7 \cdot zH_2O$  or sodium hexameta phosphates,  $Na_6P_6O_{18} \cdot zH_2O$ , wherein the metal pyro/polyphosphates are hydrates and may be in powder form. Sodium acid pyrophosphate is the most preferred.

The pyro/polyphosphate is added to the flame retarded molding compositions in an amount effective to increase track resistance but not in such amount that other essential properties of the molding composition are substantially degraded. Preferably, the final composition comprises from about 0.5 to about 40, and more preferably from about 0.5 to about 20 percent of the pyro/polyphosphate for improving the CTI.

Additionally, it has been found that olefinic type polymers may be added to the composition to enhance the CTI. Typical olefinic type polymers are the addition polymers of olefins. Suitable olefins include ethylene, propylene, butylene, hexene and octene or their copolymers.

Polyolefins may be high density polyethylene (d. above 0.94), low density polyethylene (d. about 0.92), linear low density polyethylene (d. 0.916-0.940) or polypropylene. The olefin polymers may be copolymers, including ethylene-propylene or ethylene-butylene copolymers and those containing acid or ester structural units derived, for instance, from acrylic acid, methacrylic acid, acrylic esters, methacrylic esters, alkyl vinyl esters or the like, most often in the amount of about 5-25%, preferably 10-25 percent by weight. Most preferred are copolymers of ethylene with ethyl acrylate or methacrylate.

Preferably, olefinic type polymers are included in the final composition in an amount from about 1 to about 10, and more preferably from about 3 to about 7 percent by weight of the final composition as an effective amount to improve the CTI.

Included in the resin compositions of the present invention are mineral filler materials which act with the pyro/polyphosphates to enhance the track resistance. The preferred mineral fillers enhance the CTI. Typical mineral fillers include metal sulfates, micas, clays such as aluminosilicates and aluminosilicates, talcs, glass flake, wollastonite, metal oxides such as titanium dioxide, zinc sulfide, ground quartz, and the like. Preferred mineral fillers are the talc, clays and metal sulfates. Typical silicates include calcium silicates such as Wollastonite, aluminum silicates such as kaolin which is a clay material composed of fine-grained mineral kaolinite, hydrated magnesium silicates known as talc, silica and alumina type mineral materials, and mica which is anhydrous aluminosilicate material. Clay is a hydrated aluminum silicate generalized by the formula:  $Al_2O_3 \cdot SiO_2 \cdot xH_2O$ . Talc is a preferred mineral filler.

The metal sulfate salts as well as their hydrates are a preferred mineral filler. Preferred metal sulfate salts are the Group IA and Group IIA metal sulfates with barium, calcium and magnesium sulfates being preferred.

Barium sulfate which is non-toxic and insoluble in dilute acids is especially preferred. Barium sulfate may be in the form of the naturally occurring barytes or as synthetically derived barium sulfate using well known synthetic techniques. The particle size may vary from 0.5 to 50 microns, preferably from 1 to 15 microns and most preferably 5 microns.

In most applications, the pyro/polyphosphates, mineral filler and reinforcement material such as glass fibers, comprise the filler material, and desirably represent 5 to 70%, and preferably from 10 to 50% of the total weight percent of the composition. Based on the total weight of the composition, reinforcement material is preferably present in an amount from 5 to 30 percent by weight, mineral filler in an amount from 3 to 30 percent by weight and pyro/polyphosphate in an amount from 0.5 to 20 percent by weight.

The flame retarded stiffness, the reinforcement material, is desirably a glass fiber or a glass fiber fabric. As previously discussed, the inclusion of glass fibers may be extremely effective in enhancing the CTI. The glass fibers are desirably selected to enhance the CTI.

For use as reinforcement, glass fibers may be used as filamentary fibers or as glass fiber fabric. For most applications, if material to be employed for electrical uses, it is preferred to use fibrous glass fragments comprised of borosilicate glass that is relatively soda-free. This is known as "E" glass. The filaments are made by standard processes, e.g., by steam or air blowing, flame blowing and mechanical pulling. The preferred filaments for plastic reinforcement are made by mechanical pulling. The filament diameters preferably range from about 0.001 to 0.01 inches.

In preparing the molding compositions, it is convenient to use the filamentary glass in the form of chopped strands. These strands are typically about 1 ft. to about 1 1/2 ft. long. In articles molded from the composition, the strands may be even shorter lengths. The strands are typically used as reinforcement in the molding composition, and are typically present in an amount of about 5 to 20 percent by weight of the composition.

the weight of resin. A preferred range will be from about 5 to 15 percent.

Typically halogenated aromatic flame-retardants include tetrabromobiphenyls. A polycarbonate oligomer, poly-*o*-methoxyphenyl ether, brominated polystyrene, brominated BPA polyeпоxyol, brominated imides, brominated polycarbonate, poly(thaluryl acrylate), poly(thaluryl methacrylate), or mixtures thereof. Poly(*o*-methoxyphenyl acrylate) is preferred with the most preferably being poly(pentabromobenzyl acrylate). PBB-FA has been known for some time, and is a valuable flame-retardant material useful in a number of synthetic resins. PBB-FA is prepared by the polymerization of pentabromobenzyl acrylate ester (PBB-MA). The PBB-FA polymeric flame-retardant material is incorporated into the synthetic resin during processing to impart flame-retardant character.

Examples of other suitable flame retardants are brominated polystyrenes such as polydibromostyrene and polytribromostyrene, decabromobiphenyl ethane, tetrabromobiphenyl, brominated aliphatic omega-alkylene-bis-phtalimides, e.g. N-N'-ethylene-bis-tetrabromophthalimide, oligomeric brominated carbonates, especially carbonates derived from tetrabromobisphenol A, which, if desired, are end-capped with phenoxy radicals, or with brominated phenoxy radicals, or brominated epoxy resins. Other aromatic carbonate flame retardants are set forth in U.S. Patent 4,656,544 to Hepp.

The flame retardants are typically used with a synergist, particularly those antimony compounds. Such compounds are widely available or can be made in known ways. Typical inorganic synergist compounds include  $Sb_2O_3$ ,  $Sb_2S_3$ , and the like. Especially preferred is antimony trioxide ( $Sb_2O_3$ ). Synergists such as antimony oxides, are typically used at about 0.5 to 15, and more preferably from 1 to 6 percent by weight based on the weight percent of resin in the final composition.

Also, the final composition may contain polytetrafluoroethylene (PTFE) type resins or copolymers used to reduce dripping in flame-retardant thermoplastics.

Suitable polyamide components include polyamide-6, polyamide-6-6, polyamide-11, polyamide-12, polyamide-4-6, polyamide-6-10, and polyamide-6-12, as well as polyamides prepared from terephthalic acid and/or isophthalic acid and trimethylhexamethylenediamine, from adipic acid and *m*-xylylenediamine, from adipic acid, azelaic acid, 2,2-bis-(*p*-aminocyclohexyl) propane, and from terephthalic acid and 4,4'-diaminodicyclohexylmethane. Mixtures and/or copolymers of two or more of the foregoing polyamides or prepolymers thereof, respectively, are also within the scope of the present invention.

Furthermore, the polyamides may be made by any known method, including the polymerization of a monoamino monocarboxylic acid or a lactam thereof having at least 2 carbon atoms between the amino and carboxylic acid group, of substantially equimolar proportions of a diamine which contains at least 2 carbon atoms between the amino groups and a dicarboxylic acid, or of a monoaminocarboxylic acid or a lactam thereof as defined above, together with substantially equimolar proportions of a diamine and a dicarboxylic acid. The dicarboxylic acid may be used in the form of a functional derivative thereof, for example, a salt, an ester or acid chloride.

A detailed description of polyamides and polyamide precursor materials is provided in U.S. Pat. No. 4,755,566 to Yates. Other useful polyamides often referred to as "Nylons" are disclosed in U.S. Pat. Nos. 4,732,935 to Grant et al., 4,659,760 to Van der Meer, and 4,315,086 to Leno et al., each also incorporated herein by reference. The polyimide used may also be one or more of those referred to as "toughened nylons", which are often prepared by blending one or more polyamides with one or more polymeric or copolymeric elastomeric toughening agents. Examples of these types of materials are given in U.S. Pat. Nos. 4,174,351, 4,474,927, 4,348,164, 4,251,644, 3,884,882, 4,147,740, all incorporated herein by reference, as well as in a publication by Gallucci et al., "Preparation and Reactions of Epoxy-Modified Polyethylene", J. APPL. POLYM. SCI., V27, PP. 425-437 (1982).

The preferred polyamides for this invention are polyamide-6, 6-6, 11 and 12, with the most preferred being polyamide-6-6.

The polyamides used herein preferably have an intrinsic viscosity of from about 0.4 to about 0.6 dl/g as measured in a 90/10 chloroform mixture at 25°C to 30°C.

Blends of various polyamide resins as the polyamide component, and of from about 1 to about 100 parts by weight of preferred polyamides as set forth above and from about 99 to about 1 part by weight of other polyamides based on 100 parts by weight of both components combined.

Additional ingredients may include other thermoplastic resins, in an amount up to about 50 percent by weight based on the weight of the formulation. Such other suitable thermoplastic resins which may be used include polyesters, acrylic and methacrylic polymers or copolymers, epoxy resins, poly carbonates, polyacrylonitriles, polyurethane oxide based resins, such as polyphenylene oxide and blends of polyphenylene oxide and polyethylene oxide, polyarylethers, polyphenylene ethers, polyetherketones, aliphatic polyketones, polysulfones, and epoxies, and the like, and the like.

Other ingredients employed in low amounts, typically less than 5 percent by weight of the total composition, include stabilizers, lubricants, colorants, plasticizers, nucleants, antioxidants and UV absorbers. These ingredients should be selected so as not to deleteriously affect the desired properties of the micro-bead.

Although it is not essential, best results are obtained if the ingredients are precompounded, pelletized and then molded. If recompounding can be carried out in conventional equipment. For example, after precrying the polyamide resin, other ingredients (and, optionally, additives and/or reinforcement) can be fed into a single screw extruder. A dry blend of the composition. On the other hand, a twin screw extrusion machine can be fed with resins and additives at the feed port and reinforcement downstream.

Portions of the blend can be precompounded and then extruded with the remainder of the formulation, and cut or chopped into the final compounds, such as conventional granules, pellets, etc., by standard techniques.

The compositions can be molded in any equipment conventionally used for thermoplastic compositions. For example, good results will be obtained in an injection molding machine of the 50-ton Van Dorn type, with conventional temperatures which depend on the particular thermoplastic utilized. If necessary, depending on the molding properties of the polyamide, the amount of additives and/or reinforcing filler, and the rate of crystallization of the polyamide component, those skilled in the art will be able to make conventional adjustments in molding cycles to accommodate the composition.

Flammability tests were performed with the procedure of Underwriters' Laboratories Bulletin 94 entitled "Combustion Tests for Classification of Materials - UL-94." According to this procedure, the materials were classified as either UL-94 V-0, UL-94 V-1 or UL-94 V-2 on the basis of the tests results obtained for the samples. The criteria for each of these flammability classifications according to UL-94 are briefly as follows:

V.0: the average period of flaming and/or smoldering after removing the lighting flame should not exceed five seconds and none of the samples should produce drops of particles which ignite absorbent cotton.

V-1 The average period of flaming and/or smoldering after removing the flaming flame should not exceed twenty-five seconds and none of the samples should produce drips of particles which ignite absorbent cotton.

V-2: the average period of flaming and/or smoldering after removing the flaming flame should not exceed twenty-five seconds and the samples may produce drops of burning particles with white absorbent cotton.

The following examples illustrate the preparation of certain compounds falling within the scope of this invention. They are not to be construed to limit the invention in any manner whatsoever. All parts, except as otherwise indicated, are by weight.

## Examples

The formulations shown below in Tables 1 were preblended and extruded in a 2.5-inch 30:1 L/D HPM Single Screw Extruder. The extrudate was cooled through a water bath prior to pelletizing. Test parts were injection molded in a 60-ton Van Dorn molding machine. The pellets were dried for 3-4 hours at 250°C in a forced air circulating oven prior to injection molding.

The control experiments labeled A-E showed low CTI. Compared to the control experiments, the formulations of the invention, labeled 1-7, containing pyro/polyphosphates and sulfates or minerals showed improved CTI. The pyro/polyphosphates are meta pyrophosphates, meta polyphosphates, meta acid polyphosphates, or meta acid polyphosphates.

Emulsion A was polyvinylidene fluoride (Gardolene 1000) and styrene copolymer and emulsion B was polyvinylidene fluoride (Gardolene 1000) and styrene copolymer and acrylonitrile-butadiene copolymer and emulsion C was polyvinylidene fluoride (Gardolene 1000) and styrene copolymer and acrylonitrile-butadiene copolymer and methyl methacrylate (MMA) (20 wt %).

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Combination 1 was Combination 3 with 4% less tail and replaced by sodium acid pyrophosphate (SAPP). Tail and SAPP combination showed beneficial effects on the CT, which was improved by 75 units.

Formulation 2 was Formulation 1 with 4.5% less talc and replaced by ethylhexylglycidyl ether (EEA). Talc and EEA combination showed beneficial effects on the CTI which was improved to 4000 V.

Formulation 3 was similar to Formulation 1 with 4.5% (w/w) added ethyl oleate (EO), triethyl citrate (TEC), SAPP, and ELA to treat or soothe beard hair effects on the FT, which was improved 14.5%.

Formulation C was polyamide 6 resin plus 30 percent glass plus 10 percent wax and 10 percent ethylene copolymer and 10 percent talc.

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Formulation 5 was similar to Formulation 4 with 4.5% added ethylene methacrylate (EEA). Talc, SAPP, and EEA combination showed beneficial effects on the CTI which was improved to 420 volts.

Formulation 6 was polyamide 6 resin plus 30 percent glass plus flame retardant without ethylene copolymer and fillers. The flame retardant system was bromo epoxy with antimony synergist. The CTI was 300 volts.

Formulation 7 was Formulation 6 with 10% less glass and 25% talc. It showed beneficial effects on the CTI which was improved to 375 volts.

Formulation 8 was Formulation 6 with 4% less talc and replaced by sodium acid pyrophosphate (SAPP). Talc and SAPP combination showed beneficial effects on the CTI which was improved to 425 volts.

Formulation 9 was similar to Formulation 6 with 4.5% added ethylene methacrylate (EEA). Talc, SAPP, and EEA combination showed beneficial effects on the CTI which was improved to 420 volts.

Articles which are molded from the formulated resins are suitable for use in electrical devices such as electrical outlets, circuit breakers, switches, fuse holders, lamp sockets, distributor caps, enclosures, or ignition coils. Such articles may be formed by conventional molding techniques.

TABLE 1

Material	A	B	1	2	3	C	D	4	5	E	F	6	7
SIN													
lyamide 6,6 - (1)	58.85	43.85	43.85	43.85	38.85	58.85	43.85	43.85	38.85	56.60	41.60	41.60	36.60
lyamide 6 - (2)													
ASS													
lass - (3)	30.00	20.00	20.00	20.00	20.00	30.00	20.00	20.00	20.00	30.00	20.00	20.00	20.00
*SYNERGIST													
Acrylate - (4)	8.00	8.00	8.00	8.00	8.00	8.00	8.00	8.00	8.00				
EO3 Conc - (5)	3.00	3.00	3.00	3.00	3.50	3.00	3.00	3.00	3.00	13.25	13.25	13.25	13.25
Epoxy+Sb2O3 Conc - (6)													
total		25.00	21.00	20.50	21.00		25.00	21.00	21.00		25.00	21.00	21.50
pp (7)			4.00		4.00			4.00	4.00			4.00	4.00
res													
A - (8)	0.15	0.15	0.15	4.50	4.50	0.15	0.15	0.15	4.50	0.15	0.15	0.15	4.50
max TM 1076 - (9)				0.15	0.15				0.15				0.15
	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00	100.00
4 @ 1.6mm	V-2	V-2	V-1	V-1	V-1	V-2	V-2	V-1	V-1	V-2	V-2	V-1	V-1
Volts	275	350	375	425	475	300	4.00	475	500	300	375	125	475

olyamide 6,6 resin

olyamide 6 resin

4 micron diameter glass

oly(pentabromobenzyl Acrylate) Mw 15,000

-5% Sb2O3 and 15% olefinic binder

5% Brominated Epoxy and 25% Sb2O3 concentrate

odium Acid Pyrophosphate

glycopolymer of Ethylene with 18% Ethyl Acrylate

indered Phenol Anti Oxidant

## Claims

1. A flame-retarded resin molding composition with enhanced electrical properties comprising a polyamide resin, a flame-retarding amount of a halogenated flame retardant, fiber glass and/or mineral filler, and an effective amount of a pyro polyphosphate selected from the group consisting of metal pyro polyphosphates, metal polyphosphates, metal acid pyrophosphates, metal acid polyphosphates and mixtures thereof.
2. A flame-retarded resin molding composition according to claim 1 wherein said pyro polyphosphate has the formula  $M_xH_yF_nO_{3n+1}$  wherein M is a metal, x is from 1 to 5, y is a number from 1 to 12, z is a number from 0 to 12, n is a number from 2 to 10.
3. A flame-retarded resin molding composition according to claim 2 wherein the C.T. of the final composition is greater than at out 400 volts.
4. A flame-retarded resin molding composition according to claim 1 wherein said flame retardancy is V-0 according to Underwriters Laboratories Test No. 94.
5. A flame-retarded resin molding composition according to claim 1 wherein said flame retardant is a poly (haloaryl-methacrylate), halogenated polystyrene or a poly (haloarylacrylate) flame retardant.
6. A flame-retarded resin molding composition according to claim 1 wherein said resin molding composition includes a reinforcing material.
7. A flame-retarded resin molding composition according to claim 6 wherein said pyro polyphosphate, said mineral filler, and reinforcing material such as glass fiber comprise the filler content of said resin and said filler content comprises from 10 to 50% of the total weight percent of the composition.
8. A flame-retarded resin molding composition according to claim 7 wherein based on the total weight of the composition, said glass fiber is present in an amount from 5 to 30 percent by weight, mineral filler is present in an amount from 3 to 30 percent by weight and said pyro polyphosphate is present in an amount from 0.5 to 20 percent by weight wherein said weight percents are based on the total weight of said composition.
9. A flame-retarded resin molding composition according to claim 1 additionally including an effective amount of an olefinic polymer for improving electrical properties.
10. An article molded from the flame-retarded resin molding composition according to claim 1.



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# EUROPEAN SEARCH REPORT

Application Number  
EP 85 30 0624

## DOCUMENTS CONSIDERED TO BE RELEVANT

Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to	CLASSIFICATION OF THE APPLICATION (Int. Cl. 6)
A	FR 1 545 754 A (URGESS) 7 October 1968 ---		C08K3/00 C08K3/32
A	FR 1 460 674 A (FIRESTONE) 9 February 1967 ---		
A	FR 1 464 645 A (FIRESTONE) 16 March 1967 ---		
A	EP 0 183 195 A (BASF AG) 4 June 1986 -----		
			TECHNICAL FIELDS SEARCHED (Int. Cl. 6)
			C08K

The present search report has been drawn up for all claims

Place of search	Date of completion of the search	Searcher
THE HAGUE	29 May 1996	Lendy, A
<p>ABSTRACT OF THE INVENTION</p> <p>1. The present invention relates to a process for the preparation of a composition which is particularly relevant if taken alone.</p> <p>2. The present invention also relates to a composition which is particularly relevant if combined with another composition in the same field.</p>		